

(–)-Sparteine-Mediated Stereoselective Intramolecular Carbolithiation of Alkynes

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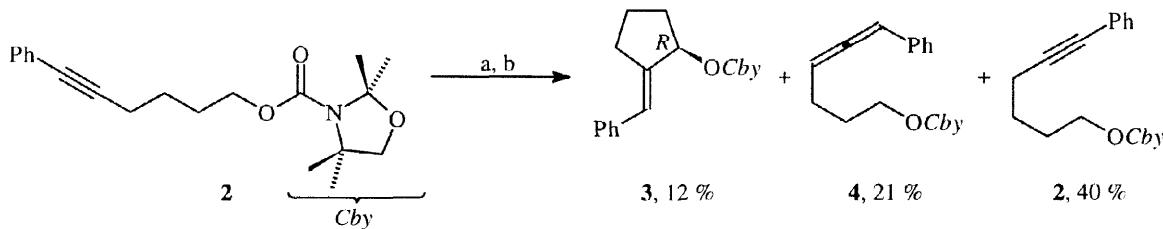
Key words: *asymmetric deprotonation, (–)-sparteine, intramolecular carbolithiation, alkylolithium derivatives, chiral alkylidene cyclopentanes*

Abstract: The asymmetric deprotonation mediated by the chiral base *s*-butyllithium/(–)-sparteine of 4-substituted 5-hexynyl carbamates permits the synthesis of enantioenriched carbanionic pairs which undergo a regioselective 5-*exo*-dig ring closure with the triple bond acting as an internal electrophile. The functionalized five-membered rings are formed with complete stereoselectivity in high yields.

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The asymmetric deprotonation of carbamate esters derived from primary alkanols with the chiral base *s*-butyllithium/(–)-sparteine (*s*-BuLi/1) and the subsequent stereospecific electrophilic substitution of the carbanionic intermediates by *external* electrophiles represent a powerful tool for the synthesis of enantioenriched secondary alkanols.¹ An extension of this method is the employment of carbon-carbon multiple bonds as *internal* electrophiles corresponding to an intramolecular carbolithiation. Bailey and others have already demonstrated for achiral substrates that especially double bonds² but also triple bonds³ can serve as good electrophiles. Therefore, our interest is focussed on the fusion of the concepts of the *asymmetric deprotonation* and the *intramolecular carbolithiation*. In this context, we have recently reported the first example of an enantioselective intramolecular carbolithiation starting from achiral 6-phenyl-5-hexenyl carbamates which cyclize stereoselectively in the presence of *s*-BuLi/1 in moderate yields to give substituted cyclopentanols incorporating three defined adjacent stereocenters.^{4,5}

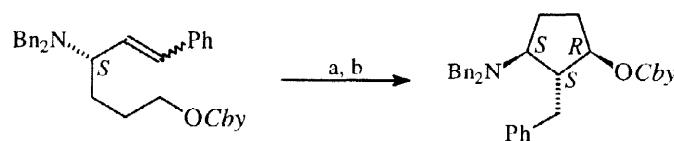
If the same concept is applied to the corresponding alkyne **2** the relatively high thermodynamic acidity of the propargylic protons competes with the kinetic acidity of the protons at the carbon bearing the activating hydroxy group. Thus, after treatment of the 6-phenyl-5-hexynyl carbamate (**2**) with *s*-BuLi/1 not only the desired cyclization product **3** but also the allene **4** were isolated in poor yields (Scheme 1).



a) 1.5 equiv. *s*-BuLi/1, Et₂O, -78 °C, 18 h; b) 2.0 equiv. MeOH, -78 °C → rt.

Scheme 1

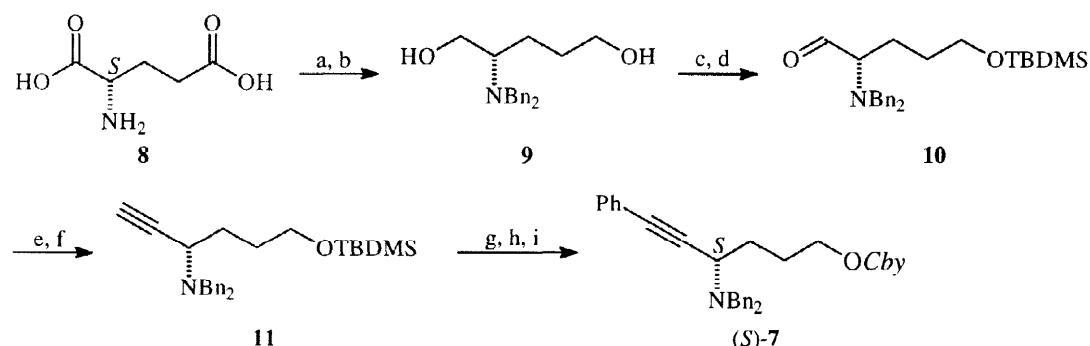
During our studies we recognized that the cyclization behaviour of 6-phenyl-5-hexenyl carbamates is dramatically enhanced if a substituent is introduced in allylic position. The chiral 4-substituted carbamate **5** derived from (*S*)-glutamic acid (**8**) cyclizes with good stereoselectivity yielding **6** in 70 % (Scheme 2).



a) 1.5 equiv. *s*-BuLi/**1**, Et₂O, -78 °C, 18 h; b) 2.0 equiv. MeOH, -78 °C → rt.

Scheme 2

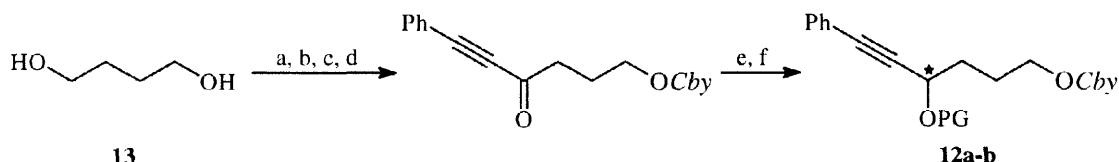
Consistently we introduced several propargylic substituents in the carbamate **2** to investigate whether this effect proves to be generally applicable to alkynes. The 4-amino-6-phenyl-5-hexynyl carbamate (*S*)-**7** was prepared starting from (*S*)-glutamic acid (**8**) using a regioselective silylation⁶ of the intermediate diol **9**. The terminally substituted triple bond was introduced by Corey's formyl→ethynyl conversion⁷ (**10**→**11**) with a subsequent sp-sp² coupling reaction known as the Sonogashira reaction⁸ (**Scheme 3**).



a) 4.5 equiv. BnBr, 2.25 equiv. K₂CO₃, 2.25 equiv. NaOH, MeOH/H₂O (1:1), reflux, 80 %; b) 1.6 equiv. LiAlH₄, THF, reflux, 97 %; c) 1.2 equiv. TBDMSCl, 0.5 equiv. DMAP, 1.0 equiv. Et₃N, CH₂Cl₂, reflux, 76 %; d) Swern, 90 %; e) 2.0 equiv. CBr₄, 4.0 equiv. PPh₃, CH₂Cl₂, 0 °C, 69 %; f) 2.0 equiv. *n*-BuLi, -78 °C → rt, 85 %; g) 1.0 equiv. PhI, 1.0 mol% (Ph₃P)₂PdCl₂, 0.5 mol% CuI, Et₃N, rt, 94 %; h) 3.0 equiv. TBAF, THF, rt, 94 %; i) 1.2 equiv. CbyCl, 1.2 equiv. NaH, THF, reflux, 88 %.

Scheme 3

The synthesis of several 4-hydroxy-6-phenyl-5-hexynyl carbamates **12a-b** was accomplished in a straightforward manner starting from 1,4-butanediol **13** with Brown's Alpine-Borane® reduction⁹ as the key step; both enantiomers of **12** could be synthesized with an enantiomeric excess of 90 %, determined via esterification with Mosher's reagent¹⁰ (**Scheme 4**).

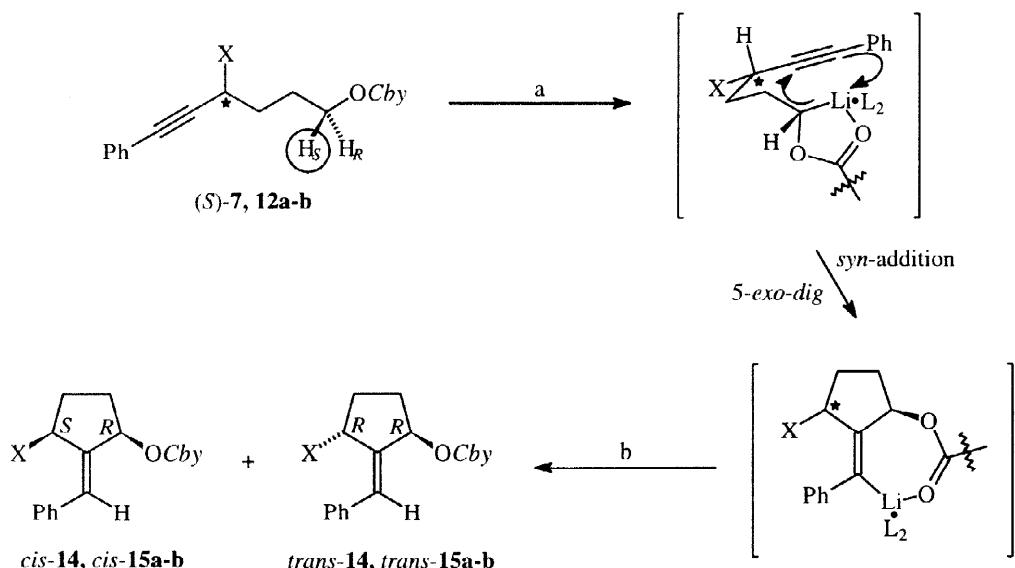


a) 0.33 equiv. CbyCl, 0.35 equiv. NaH, THF, reflux, 87 %; b) Swern, 89 %; c) 1.3 equiv. phenylacetylene, 1.25 equiv. *n*-BuLi, THF, -78 °C → -20 °C → rt, 95 %; d) MnO₂, CH₂Cl₂, reflux, 82 %; e) 2.0 equiv. Alpine-Borane®, acetaldehyde, 2.2 equiv. ethanolamine, no solvent, rt, 87 %; f) PG = Tr: 1.1 equiv. TrCl, 1.5 equiv. Et₃N, 0.05 equiv. DMAP, CH₂Cl₂, reflux, 54 %; PG = TBDPS: 1.1 equiv. TBDPSCl, 1.1 equiv. Et₃N, 0.5 equiv. DMAP, CH₂Cl₂, reflux, 98 %.

Scheme 4

When the precursor (*S*)-**7** was analogously treated with *s*-BuLi/**1** in Et₂O the desired cyclization product *cis*-**14**^{11a} could be isolated in 70 % yield (Table 1, entry 1), again showing the enhancing effect of a substituent in

the 4-position of 5,6-unsaturated alkenyl and alkynyl carbamates (Schemes 2 and 5); the formation of the allene was not observed. The other carbamates (*S*)-**12a** and (*S*)-**12b** also reveal this smooth cyclization behaviour undergoing the ring closure nearly quantitatively to yield the functionalized cyclopentylidene derivatives *cis*-**15a** and *cis*-**15b** (Table 1, entries 2 and 5); a crystal structure of *cis*-**15a**¹² allowed the determination of the relative configuration and proved that the chiral base *s*-BuLi/**1** selectively removes the *pro-S*-proton in (*S*)-**7** and **12a-b**. The *cis:trans* ratios of the cyclization products obviously correspond to the enantiomeric excess in the particular precursor; thus, the new stereocenter is formed highly stereoselectively.



a) 1.5 equiv. *s*-BuLi/L₂ (L₂ = (-)-sparteine (**1**) or TMEDA (**16**)[a]), Et₂O, -78 °C, 18-22 h; b) 2.0 equiv. MeOH, -78 °C → rt.

Scheme 5

In order to investigate the role of the existing stereocenter the (*R*)-configured precursor (*R*)-**12a** and the racemates *rac*-**12a** and *rac*-**12b** were cyclized by the typical procedure (Table 1, entries 3, 4 and 6). These experiments furnished the cyclization products *trans*-**15a**, **15a** and **15b** exclusively in *cis:trans* ratios which are directly related to the ratio of the enantiomers of compounds (*R*)-**12a**, *rac*-**12a** and *rac*-**12b**. The fact that the cyclizations of the racemic carbamates *rac*-**12a** and *rac*-**12b** gave the products with *cis:trans* ratios of 50:50 in all cases exhibits that there is no kinetic resolution of the enantiomers operating (Table 1, entries 4 and 6).

Table 1: Stereoselective Cyclization of the Precursors (*S*)-**7** and **12a-b**¹¹

entry	precursor	X	ratio of the enantiomers	configuration at C★	diamine	major product	<i>cis:trans</i> ratio	yield (%)
1	(<i>S</i>)- 7	NBn ₂	>99 : 1	<i>S</i>	1	cis-14	>99 : 1	70
2	(<i>S</i>)- 12a	OTr	95 : 5	<i>S</i>	1	cis-15a	95 : 5	88
3	(<i>R</i>)- 12a	OTr	5 : 95	<i>R</i>	1	trans-15a	5 : 95	99
4	<i>rac</i> - 12a	OTr	50 : 50	<i>R, S</i>	1	15a	50 : 50	80
5	(<i>S</i>)- 12b	OTBDPS	95 : 5	<i>S</i>	1	cis-15b	95 : 5	90
6	<i>rac</i> - 12b	OTBDPS	50 : 50	<i>R, S</i>	1	15b	50 : 50	96
7	(<i>S</i>)- 12b	OTBDPS	95 : 5	<i>S</i>	16	<i>epi</i> - 15b ^[a]	50 : 50	89

[a] In this case the diastereomers *cis*-**15b** (1*R*,3*S*) and *ent-trans*-**15b** (1*S*,3*S*) are formed with an enantiomeric excess of 90 %.

If the cyclization precursor (*S*)-**12b** is treated with *s*-BuLi in the presence of an achiral diamine such as TMEDA (**16**), the formation of the new stereocenter proceeds not being affected by the existing one; compound *epi*-**15b** is obtained in high yield but with a *cis:trans* ratio of 50:50 (Table 1, entry 7).

In summary, we have shown that the intramolecular addition of a chiral carbanion to a triple bond occurs in a *syn*-fashion and is completely regioselective; the 5-*exo-dig* cyclization product is formed exclusively. This method represents an extension to the intramolecular carbolithiation of alkenes⁴ allowing the stereoselective synthesis of substituted enantiopure cyclopentanoid building blocks.

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- 10. (a) Dale, J. A.; Dull, D. L.; Mosher, H. S. *J. Org. Chem.* **1969**, *34*, 2543-2549; (b) The secondary alkanols were reacted with (*R*)-(–)- α -methoxy- α -(trifluoromethyl)-phenylacetic acid chloride, and the enantiomeric excess was determined from the ¹H- and ¹⁹F-NMR.
- 11. (a) Compound *cis*-**14**: ¹H-NMR (600 MHz, CDCl₃): δ (ppm) = 1.33-1.60 (7s, 12H, 4CH₃ (*Cby*)); 1.78-1.85 (m, 1H, 5-H α); 1.96-2.04 (m, 2H, 4-H α , 5-H β); 2.18-2.28 (m, 1H, 4-H β); 3.24 (2d, 2H, 2J = 12.6 Hz, NCH₂); 3.70 (2s, 2H, CH₂ (*Cby*)); 3.85 (m, 2H, NCH₂); 4.20 (m, 1H, 3-H); 5.56 (m, 1H, 1-H); 6.92 (m, 1H, 6-H); 7.02-7.04 (m, 3H, Ph); 7.16-7.40 (m, 12H, Ph); ¹³C-NMR (150 MHz, CDCl₃): δ (ppm) = 24.1/24.2/25.2/25.3/25.4/25.5/26.6/26.7 (4CH₃ (*Cby*)); 22.1 (C-4); 30.3/30.4 (C-5); 54.8/54.9 (NCH₂); 59.5/60.8 (NC(CH₃)₂ (*Cby*)); 60.4 (C-3); 76.0/76.4 (OCH₂ (*Cby*)); 80.3 (C-1); 94.6/96.0 (NC(CH₃)₂O (*Cby*)); 126.8/127.1/127.6/127.9/128.3/128.7/129.3/130.3/135.9/136.0/139.2 (Ph, C-2); 141.7/141.8 (C-6); 151.9/152.6 (C=O); $[\alpha]^{25}_D$ = -76.6 (c = 0.15, CHCl₃); (b) All isolated compounds gave satisfactory analytical and spectroscopic data.
- 12. Crystal structure data of *cis*-**15a**: Crystallographic data (excluding structure factors) for the structures reported in this paper have been deposited with the Cambridge Crystallographic Data Centre.